

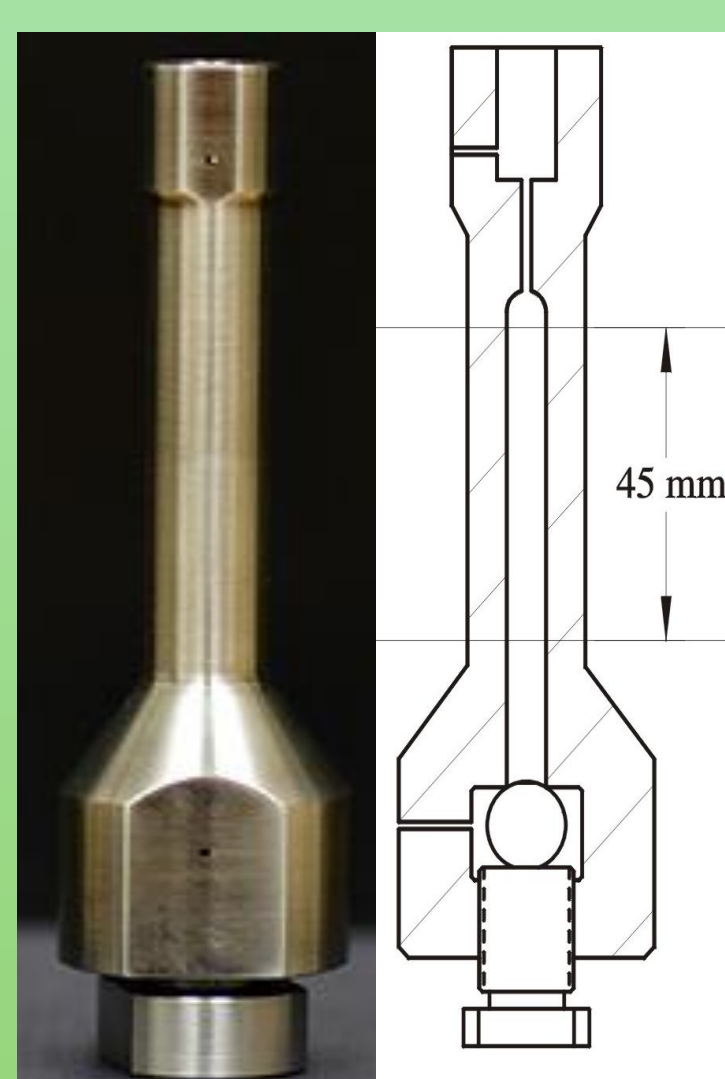
Hydrostatic High Pressure Capabilities at the NIST Center for Neutron Research

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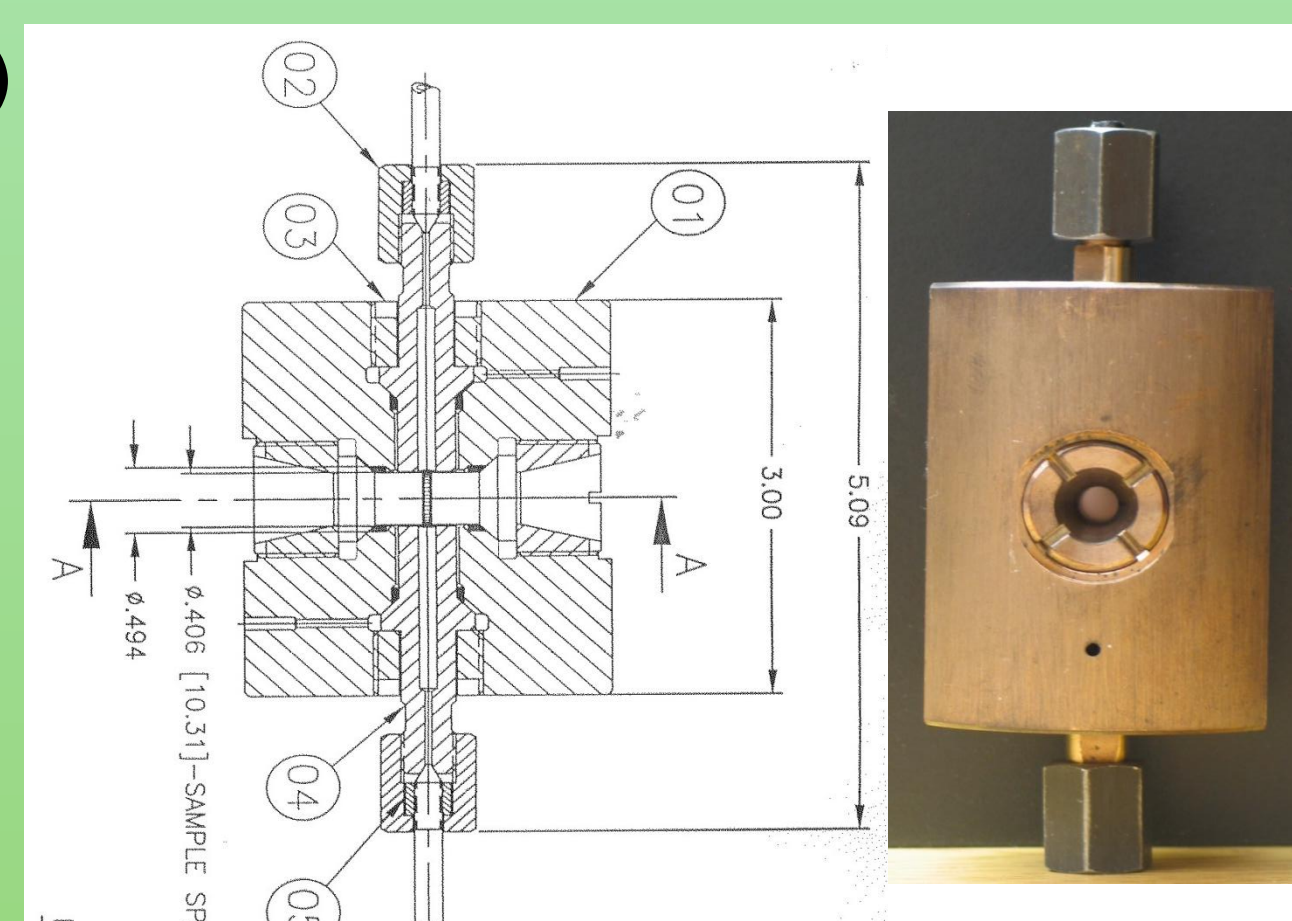
The NIST Center for Neutron Research currently provides a series of pressure apparatus ranging to as high as 2.5 GPa that are specially designed for neutron spectroscopy. Most of the pressure equipment can be mounted in a variety of instruments throughout NCNR's facility, allowing for experimental flexibility and maximizing beam time use.

(a) $P_{max} = 1.0 \text{ GPa}$
 (c) $1.5\text{K} < T < 350\text{K}$
 13-8Mo stainless steel construction
 (No cobalt)
 2.2 cm³ total sample volume
 1.5 cm³ effective illuminated volume
 25% neutron transmission at 2Å
 Inert gases



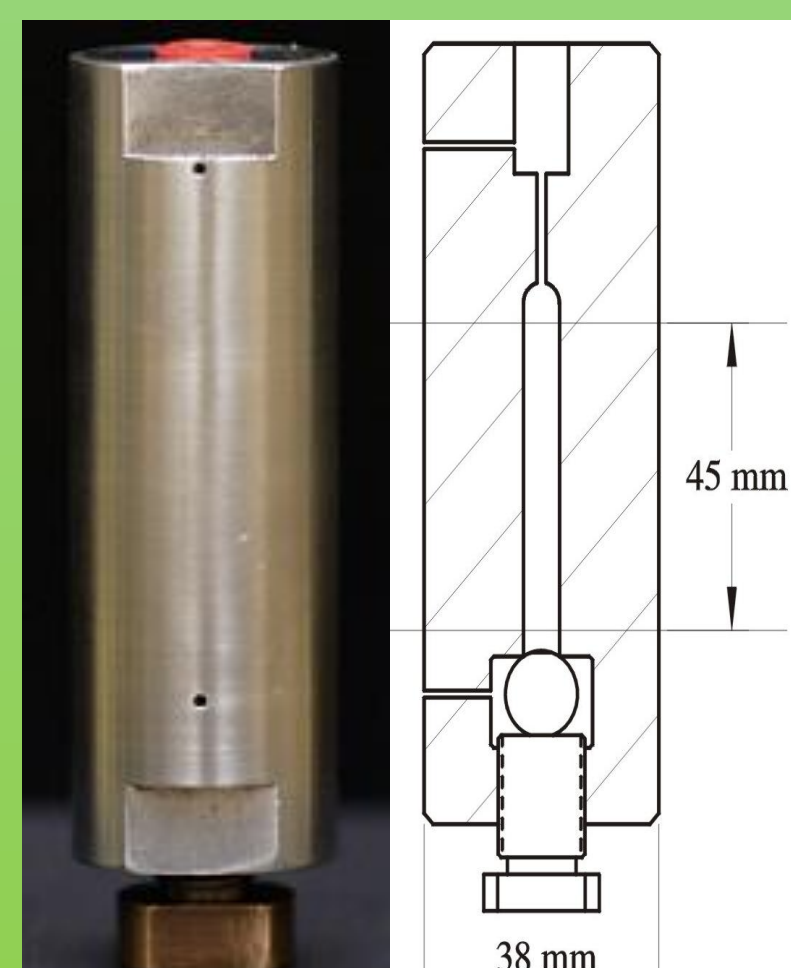
1.0 GPa Inert Gas Pressure Vessel

(b) $P_{max} = 300 \text{ MPa}$
 $\text{LN}_2 < T < 350\text{K}$
 CuBe construction with sapphire windows
 2.2 cm³ total sample volume
 Neutron beam cross-section area 0.3 cm²
 Beam divergence angle $\theta \approx 15^\circ$
 Inert gases



300 MPa Sapphire Window Pressure Vessel

$P_{max} = 650 \text{ MPa}$
 $1.5\text{K} < T < 300\text{K}$
 Aluminum 7075-T6 construction
 1.5 cm³ sample volume
 .635 cm dia. x 5.08 cm sample illumination
 60% Neutron Transmission at 2Å
 Inert Gases



650 MPa Inert Gas Pressure Vessel

(a) $P_{max} = 200 \text{ MPa}$
 (b) $\text{LN}_2 < T < 350\text{K}$
 (e) S.S. Construction with sapphire windows
 1.0 Cm² total sample cross section area
 Adjustable sample thickness
 Beam divergence angle $\theta \approx 20^\circ$
 Methane, CO₂, H₂, inert gases, fluids



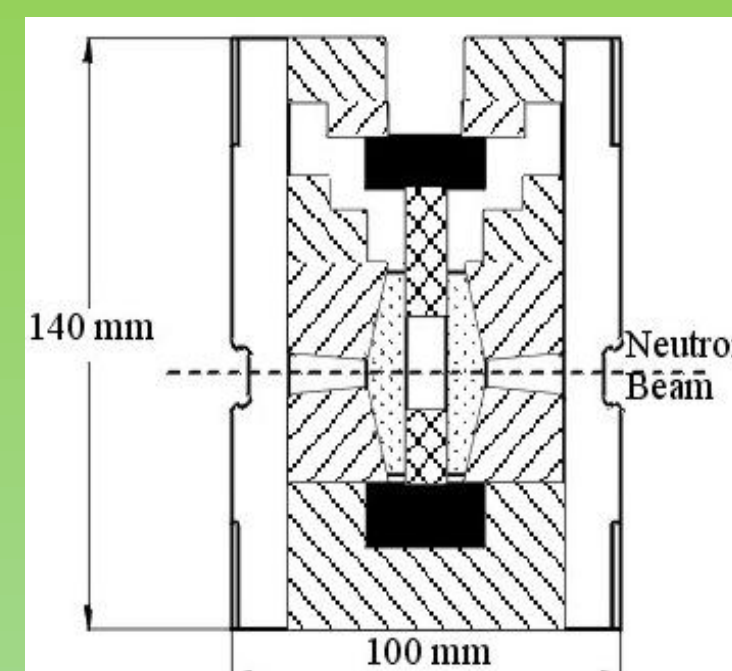
200 MPa Sapphire Window SANS Pressure Vessel

(d) Air sensitive/gas loading
 $P_{max} = 1.4 \text{ MPa (V)}$
 Up to 0.50 MPa (al)
 $1.5\text{k} \leq T \leq 800\text{k}$
 Heated gas line available for methane and CO₂



Air sensitive closure with vanadium sample can

$P_{max} = 2.5 \text{ GPa}$
 $1.5\text{K} \leq T \leq 300\text{K}$
 Sample Size:
 10 mm x 6 mm Ø



(a)



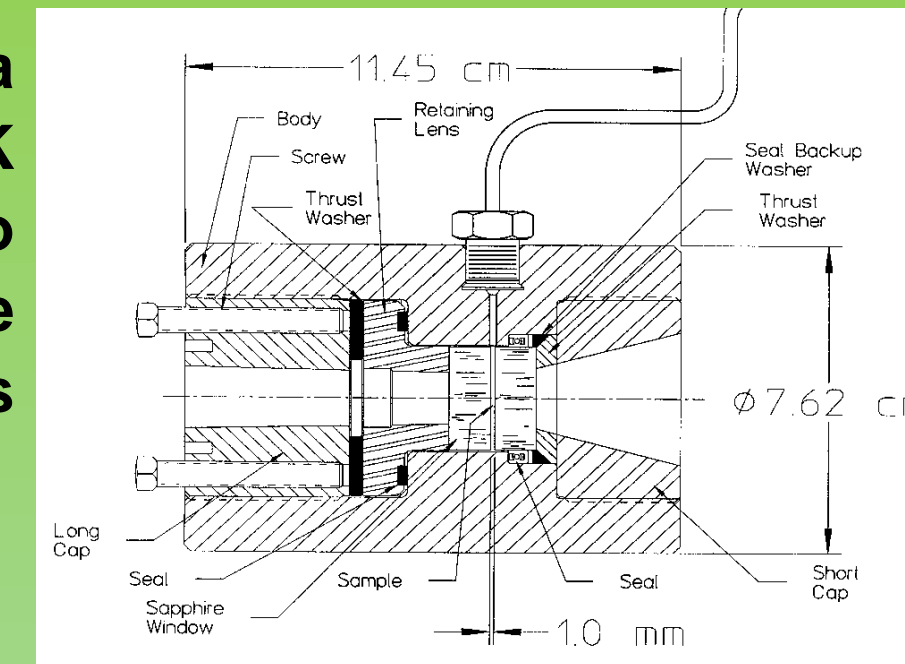
McWhan Clamp Cell schematics and as seen mounted on SD-55 (below)

The sample is axially pressurized between two opposing cylindrical Tungsten Carbide pistons. Hydrostatic uniformity is ensured by immersing the sample in a pressure transmitting media such as Fluorinert (C₈F₁₈), or a 4:1 mixture of deuterated methanol and ethanol. Pressure is monitored through the change in lattice parameters of NaCl

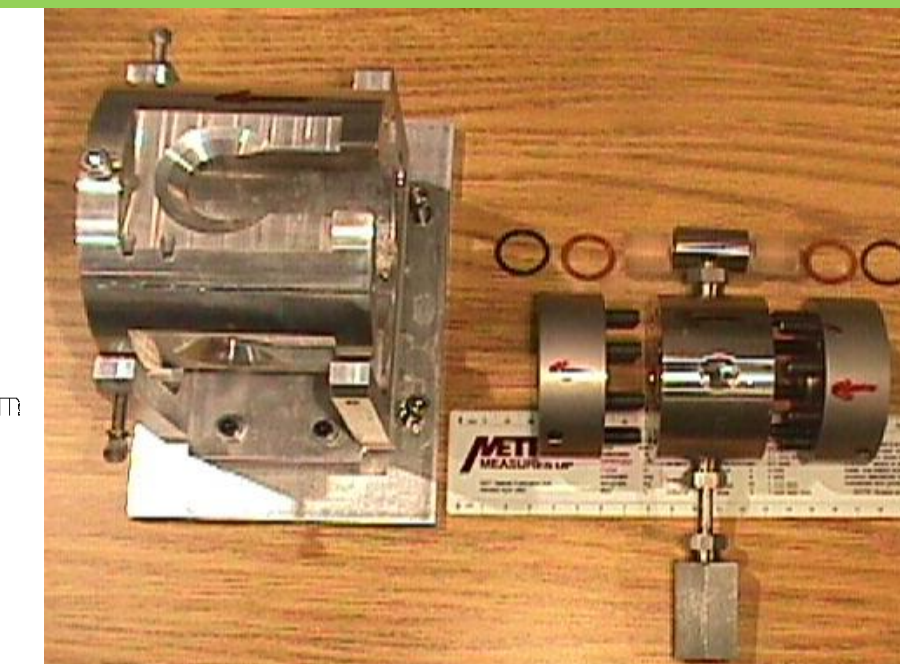
Partial list of Citations:

- T. Hong, et al. "Neutron scattering study of a quasi-2D spin-1/2 dimer system Piperazinium Hexachlorodicuprate under hydrostatic pressure." IN PRESS.
- C. Tsao, et al. "Neutron Scattering Methodology for Absolute Measurement of Room-Temperature Hydrogen Storage Capacity and Evidence for Spillover Effect in a Pt-Doped Activated Carbon." *J. Phys. Chem. Lett.* 2010, 1, 1569–1573.
- N. P. Butch, et al. "Antiferromagnetic critical pressure in URu₂Si₂ under hydrostatic conditions." *Phys. Rev. B* 82, 060408 R 2010.
- C.M. Brown, et al. "Hydrogen adsorption in HKUST-1: a combined inelastic neutron scattering and first-principles study", *Nanotechnology* 20, 204025 (2009).
- A. Kreyssig, et al. "Pressure-induced volume-collapsed tetragonal phase of CaFe₂As₂ as seen via neutron scattering." *Physical Review B* 78, 184517 (2008)
- A.J. Patel, et al. "Observing Nucleation Close to the Binodal by Perturbing Metastable Polymer Blends." *Macromol.* 40(5), 1675 (2007).

$P_{max} = 300 \text{ MPa}$
 $258\text{K} \leq T \leq 440\text{K}$
 S.S. body with two containment sapphire windows



SANS Hydraulic Pressure Vessel



(f)

This vessel can be used to pressurize polymer melt mixtures in a wafer form (blends and copolymers confined by an encapsulated o-ring) or soft macromolecular fluids (solutions, micellar systems). Also used to pressurize flowing liquids through the use of a separator (high pressure tubing containing a piston between the pressurizing fluid and the sample).